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CHEMISTRY IX NOTES ELECTROCHEMISTRY – MULTIPLE CHOICE QUESTIONS

9. The unit of electro chemical equivalent is  
a. Ampere b. Coulomb  
c. G/coulomb d. V kg/coulomb

10. In dry cell electrolyte is the moist paste of  
a.  $\text{NH}_4\text{Cl} + \text{MnO}_2$  b.  $\text{MnO}_2 + \text{ZnCl}_2$   
c.  $\text{NH}_4\text{Cl} + \text{ZnCl}_2$  d.  $\text{NH}_4\text{Cl} + \text{ZnSO}_4$

11. A weak electrolyte is:  
a. Sulphuric b. Nitric acid  
c. A caustic alkali d. Citric acid

12. The electrolysis of fused sodium chloride is carried out in:  
a. Galvanic cell b. Voltaic cell  
c. Down's cell d. Dry cell

13. The electrodes used in the electrolytic cell to carry out electrolysis of water are made of:  
a. V Platinum b. Carbon  
c. Graphite d. Copper

14. During electrolysis of which hydrogen is released at the cathode and oxygen is released at the anode?  
a. V Water b. NaCl  
c. HgO d.  $\text{H}_2\text{O}_2$

15. One Faraday has  
a. V 96,500 C b. 94,500 C  
c. 93,9500 C d. 92,500 C

16. The deposition of a metal on another metal is called:  
a. Electrolysis b. Oxidation  
c. Reduction d. V Electroplating

17. In electroplating, the thing to be electroplated is made the:  
a. V Cathode b. Electrolytic cell  
c. Anode d. Electrolyte

18. In which cells, oxidation loss of electrons occurs at anode and reduction gain of electrons occurs at cathode?  
a. V Electrochemical b. Electrolytic  
c. Primary d. Secondary

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